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Description

The present invention relates to an anti-inflammatory analgesic adhesive preparation having excellent percutaneous absorption properties.

Non-steroidal anti-inflammatory analgesic agents do not exhibit such serious side effects as exhibited in steroidal anti-inflammatory analgesic agents, and are widely used clinically.

In steroidal anti-inflammatory analgesic agents, and are widely used clinically.

However, these non-steroidal anti-inflammatory analgesic agents have a disadvantage that they still cause various side effects such as gastro-intestinal lesions although not to the extent that is caused by

cause various side effects such as gastro-intestinal lesions although not to the extent that is caused by serioidal anti-inflammatory analgesic agents. In order to minimize such side effects, various dosage forms remove under extensive investigation.

In order to overcome the above problems of side effects and to maintain the drug effects for a long period of time, a recent administration method is for percutaneous absorption of the effective component. Various ointments and adhesive preparations containing an effective component have been device.

for use in this method.

However, the skin has a stratum corneum containing keratin as a major component and further contains a large amount of a fat-soluble component such as fat, wax and cholesterol. Therefore, the skin has a physiological defensive function, a so-called "barrier function", and as a result, it is difficult to easily cause a percutaneous absorption of a drug.

In particular, many non-steroidal anti-inflammatory analgesic agents, the utility of which is highly valued, have a salt form, and the skin exhibits a strong barrier function against drugs in a salt form.

On the other hand, skin adhesive preparations are composed of a pressure-sensitive adhesive material comprising a rubber or acrylic high molecular weight material as a base material. These materials prepared to the control of the c

As a result of extensive investigations on an adhesive preparation which overcomes the above disadvantages, and increases the solubility and percutaneous absorption of non-steroidal anti-inflammatory analgesic agent having a salt form, thereby exhibiting the effect in treating disease, it has 39 been found that if a non-steroidal anti-inflammatory analgesic agent having a salt form is contained in a pressure-sensitive adhesive material layer in combination with an organic acid, the solubility of the non-steroidal anti-inflammatory analgesic agent having a salt form in the pressure-sensitive adhesive material increases and the transfer of the drug to the skin surface is facilitated, whereby the non-steroidal anti-inflammatory analgesic agent can easily penetrate through the stratum comeum as a barrier layer.

An object of the present invention is to provide an anti-inflammatory analgesic adhesive preparation having high percutaneous absorption properties of a non-steroidal anti-inflammatory analgesic agent having a salt form.

Another object of the present invention is to provide an anti-inflammatory analgesic adhesive preparation in which a non-steroidal anti-inflammatory analgesic agent having a salt form is uniformly dissolved in a pressure-sensitive adhesive material.

The anti-inflammatory analgesic adhesive preparation of the present invention comprises a flexible support having laminated thereon a pressure-sensitive adhesive material layer containing a non-steroidal anti-inflammatory analgesic agent in a salt form (as defined below) and an organic acid.

The term "a non-steroidal anti-inflammatory analgesic agent in a salt form" as used herein excludes

Dichlofenac Sodium as disclosed in EP—A—0209975 which has been published after the priority date of the
present patent specification.

The pressure-sensitive adhesive material layer which can be used in the present invention is a layer containing and maintaining a non-steroidal anti-inflammatory analgesic agent having a salt form as offective component and an organic acid as an additive to increase solubility and percutaneous absorption of the non-steroidal anti-inflammatory analgesic agent having a salt form. The material for such layer to particularly limited so long as it is a material capable of achieving the above objects and it is a layer made of a material capable of adhering to the skin surface.

High molecular weight adhesive materials can be used as the pressure-sensitive adhesive materials. Examples of the materials are acrylic pressure-sensitive adhesive materials; rubbers such as silicone 57 rubber, polyisoprene rubber, polyisobutylene rubber, polybutadiene, styrene-butadiene (or isoprenostyrene block copolymer rubber, arrylic rubber and natural rubber; yimyl-based high molecular weight materials such as polyvinyl silkyl ether, polyvinyl acetate, a partially saponified product of polyvinyl acetate, polyvinyl alchorl and polyvinyl pyrrolidone; cellulose derivatives such as methyl cellulose, carboxymethes cellulose and hydroxypropyl cellulose; polysaccharides such as pullulan, dextrin and agar; polyurethane cellulose size in polyvinyl pyrrolidors.

Of these compounds, acrylic pressure-sensitive adhesive materials are preferred from standpoints of adhesive properties to the skin and stability of the drug, in particular, pressure-sensitive adhesive materials comprising copolymers of an skly ester of (methacytic acid, an alky) ester of (methacytic acid containing an ether bond in the molecule and copolymertable monomers other than the above-described monomers are used as materials having low skin irritating properties and solubility of the druos.

Examples of acrylic pressure-sensitive adhesive materials selected from the standpoints of adhesive properties to the skin and stability of the drug are homo- or copolymers of at least one of alkyl seters of (methlacrylic acid such as butyl (methlacrylate, pentyl (methlacrylate, hexyl (methlacrylate, hexyl (methlacrylate, hexyl (methlacrylate, out) (methlacrylate, and tridexyl (methlacrylate, and tridexyl (methlacrylate, and tridexyl methlacrylate, and copolymers of at least one of the above esters and other monomers copolymers/sable therewith.

Examples of the copolymerizable monomer include carboxyl group-containing monomers such as (meth)acrylic acid, itaconic acid, crotonic acid, maleic acid, maleic anhydride and fumaric acid; sulfoxyl group-containing monomers such as styrenesulfonic acid, arylsulfonic acid, sulfopropyl acrylate, 10 (meth)acryloyloxynaphthalenesulfonic acid, acrylamidomethylpropanesulfonic acid and acryloyloxybenzenesulfonic acid: hydroxyl group-containing monomers such as hydroxyethyl (meth)acrylate and hydroxypropyl (meth)acrylate; amide group-containing acrylic monomers such as (meth)acrylamide, dimethyl(meth)acrylamide, N - butylacrylamide, tetramethylbutylacrylamide and N - methylol(meth)acrylamide: alkylaminoalkyl group-containing acrylic monomers such as aminoethyl (meth)acrylate, 15 dimethylaminoethyl (meth)acrylate, diethylaminoethyl (meth)acrylate and tertbutyl (meth)acrylate; alkyl esters of acrylic acid containing an ether bond in the molecule thereof such as methoxyethyl (meth)acrylate, ethoxyethyl (meth)acrylate, butoxyethyl (meth)acrylate, tetrahydrofurfuryl (meth)acrylate, methoxyethylene glycol (meth)acrylate, methoxydiethylene glycol (meth)acrylate, methoxypolyethylene glycol (meth)acrylate and methoxypolypropylene glycol (meth)acrylate; vinyl monomers such as N -(meth)acryloylamino acid; functional monomers such as acrylic monomers such as urethane, urea or isocyanate ester of acrylic acid; and vinyl monomers such as (meth)acrylonitrile, vinyl acetate, vinyl propionate, vinyl pyrrolidone, vinyl pyridine, vinyl pyrazine, vinyl piperadine, vinyl piperidone, vinyl pyrimidine, vinyl pyrrole, vinyl imidazole, vinyl caprolactam, vinyl oxazole, vinyl thiazole, vinyl morpholine, styrene, a-methylstyrene and bis(N,N' - dimethylaminoethyl) maleate.

The above alkyl esters of (meth)acrylic acid and copolymerizable monomers include isomers in which the alkyl portion is straight or branched, and isomers and derivatives in which the position of substituents is different

It is desirable from a standpoint of the balance between adhesive properties to the skin and cohesion that the ratio of the alkyl ester of (meth)acrylic acid to the copolymerizable monomer in the acrylic pressure-sensitive adhesive material is 50:50 to 93:1 by weight. When alkyl esters of (meth)acrylic do containing an ether bond in the molecule thereof are used from the standpoint of the low skin irritating properties, it is desirable that the ratio of the alkyl ester of (meth)acrylic acid containing an ether bond in the molecule/the other copolymerizable monomer is 40 to 80/59 to 10/1 to 40.

When the above composition is used, in the case where there is the problem that after adhering to the skin, it causes the phenomenon of adhesive transfer on the applied skin thereby contaminating the skin surface, it is preferred that the composition is subjected to suitable chemical crosslinking treatment (e.g., copolymerization of cross-inhable monomers and addition of a crosslinking agent) or physical crosslinking treatment (e.g., irradiation with ultraviolet rays or ionizing radiations such as an electron beam) to such an extent of not deteriorating the adhesive properties to the skin.

As salts of the non-steroidal anti-inflammatory analgesic agent having a salt form which can be used in the present invention, any salts can be used so long as they are pharmaceutically acceptable. For example, alkali metal salts, alkaline earth metal salts, aluminum salts and the like are preferred. Examples thereof are the salts of indomethain, fulrenamic acid, melenamic acid,

The amount of the non-steroidal anti-inflammatory analgesic agent in salt form which is present in the pressure-sensitive adhesive material is not limited so long as the therapeutic effect is exhibited. This amount of the analgesic agent is generally 1 to 40 wt%, and preferably 5 to 30 wt%, based on the weight of the pressure-sensitive adhesive material, and 20 to 1,600 µg/cm², and preferably 100 to 1,200 µg/cm² per unit area.

Since the non-steroidal anti-inflammatory analgesic agent used in the present invention is in a salt form, it is difficult to disolvo a large amount of the non-steroidal anti-inflammatory analgesic agent in the pressure-sensitive adhesive material layer having relatively high lipophilic properties and maintain the agent therein. Even if a large amount of the non-steroidal anti-inflammatory analgesic agent agent is incorporated, in some cases all the drug cannot be dissolved or crystallization of the drug occurs, making it or mossible to diffuse a sufficient amount of the drug to the skin surface.

The present invention overcomes this problem by concurrently using an organic acid. The use of the organic acid increases the solubility of the non-steroidal anti-inflammatory analgesic agent in salt form in the pressure-sensitive adhesive material laver and also the percutaneous absorption properties.

It is believed that the reason for this is that since by concurrently using the non-steroidal anti-inflammatory analgesic agent having a salt form and the organic acid, the analgesic agent is converted

into free-based drug having higher oleophilicity, the solubility of the drug in the pressure-sensitive adhesive material layer is increased and the drug can easily penetrate through the stratum corneum having the barrier function, viz., the percutaneous absorption properties are increased.

As such organic acids, it is preferred to use acids stronger than the free-based non-steroidal 5 anti-inflammatory analgasic agent, and carboxylic acids are particularly preferred. Examples of carboxylic acids include citric acid, succinic acid, tartaric acid, maleic acid, fumaric acid, selicylic acid and acetic acid. Citric acid, succinic acid and tartaria acid are particularly preferred.

The amount of the organic acid added in the pressure-sensitive adhesive material layer is from 5 to 100 parts by weight, preferably from 10 to 50 parts by weight, per 100 parts by weight of the non-steroidal anti-inflammatory analysics agent is a salt form.

As a support on which the pressure-sensitive adhesive material layer containing the analgesic agent as lift form and organic acid is provided. a material laving flustibility is chosen in order to conform to the movement of the skin surface. Examples of the supports are a plastic film, nonwoven fabrics, woven fabrics, above, a metallic folia, i foamed film or combinations thereof.

As described above, in the anti-inflammatory analyseic adhesive preparation of the present invention, the organic acid which is compounded in the pressure-sensitive adhesive material in combination with the non-steroidal anti-inflammatory analyseic agent having a salt form which is sparingly souble in the pressure-sensitive adhesive material has a function of increasing the solubility of the drug in the pressure-sensitive adhesive material and increasing the procuratenous absorption properties of the drug.

Accordingly, in the anti-inflammatory analgesic adhesive preparation of the present invention, the analgesic agent in the adhesive is percutaneously absorbed easily, thereby effectively reating inflammation and painful diseases. Furthermore, since the preparation can be externally administered, the side effect is low and the therapeutic effect can be exhibited continuously.

The present invention is described in greater detail by reference to the following illustrative examples wherein all parts are by weight unless otherwise indicated.

Example 1

55 Parts of 2-ethylhexyl acrylate, 30 parts of methoxyethyl acrylate, 15 parts of vinyl acetate and 0.3 part of azobisisobutylonitrile were placed in a four-necked flask and the mixture was heated to a temperature of 80 to 63°C in an inert gas atmosphere to initiate the polymerization reaction. The reaction was continued for 10 hours while controlling the reaction temperature by adding dropwise 125 parts of acetate. The reaction solution was further aged 75 to 80°C for 2 hours to prepare a copolymer solution.

To the copolymer solution thus obtained were added tolmetin sodium and citric acid in such amounts: that the contents of tolmetin sodium and citric acid after drying were 20 wt% and 4 wt% based on the swelght of the pressure-sensitive adhesive material layer, respectively, and the resulting mixture was coated on a releasing liner made of a polyester in such an amount that the drug content was 400 µg/cm² and then dried to prepare a pressure-sensitive adhesive material layer.

This pressure-sensitive adhesive material layer was transferred to a nonwoven fabric with an ethylene-vinyl acetate concluder. 28 wfx) having a thickness of 40 µm laminated thereon at the ethylene-vinyl acetate concluder layer side to produce an anti-inflammatory analgesic adhesive preparation of the present invention.

Example :

The anti-inflammatory analgesic adhesive preparation was prepared in the same manner as in Example 1 except that amfenac sodium and maleic acid were added to the copolymer solution in such amounts that the contents of amfenac sodium and maleic acid after drying were 20 wt% and 4 wt%, respectively.

Example 3

95 Parts of 2-ethylhexyl acrylate, 5 parts of acrylic acid and 0.2 part of benzoyl peroxide were placed in a four-necked flask and the mixture was heated to a temperature of 62 to 65°C in an inert gas atmosphere to initiate the polymerization reaction. The reaction was continued for 8 hours while controlling the reaction temperature by adding dropwise 125 parts of ethyl acetate. The reaction solution was further aged for 2 hours at 75 to 80°C to prepare a copolymer solution.

To the copolymer solution thus obtained were added loxoprofen sodium and succinic acid in such amounts that the contents of loxoprofen sodium and succinic acid after drying were 10 wt% and 3 wt% based on the weight of the pressure-sensitive adhesive material layer, respectively. The resulting mixture was coated on a releasing liner made of a polyester in such an amount that the drug content was 400 µg/cm², and then dried to prepare a pressure-sensitive adhesive material layer.

This pressure-sensitive adhesive material layer was transferred to an ethylene-vinyl acetate copolymer film (vinyl acetate content: 28 wt%) having a thickness of 30 µm to produce an anti-inflammatory analgesic adhesive preparation of the present invention.

Example 4

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The anti-inflammatory analgesic adhesive preparation was prepared in the same manner as in

Example 3 except that sodium meclofenamate and citric acid were added to the copolymer solution in such amounts that the contents of sodium meclofenamate and citric acid after drying were 20 wt% and 6 wt%, respectively and the resulting mixture was coated on a releasing liner made of a polyester in such an amount that the dry content was 800 µg/cm² and dried to prepare a pressure-sensitive adhesive material syer.

Example 5

A mixture of 80 parts of 2-ethylhexyl acrylate and 20 parts of vinyl acetate was copolymerized in the same manner as in Example 1.

To the copolymer solution thus obtained were added fenoprofen calcium and tartaric acid in such amounts that the contents of fenoprofen calcium and tartaric acid after drying were 30 wt%, respectively. The resulting mixture was coated on a releasing liner made of a polyester in such an amount that the drug content was 600 µg/cm², and then dried to prepare a pressure-sensitive adhesive material

5 This pressure-sensitive adhesive material layer was transferred to a polyethylene film having a thickness of 30 µm to produce an anti-inflammatory analgesic adhesive preparation of the present invention.

Example 6

20 Parts of polyisobutyrene rubber (viscosity-average molecular weight: 1,200,000), 30 parts of polyisobutylene rubber (viscosity-average molecular weight: 35,000), 20 parts of polybutene and 30 parts of wood resin were dissolved in a toluene/ethyl acetate (volume ratio: 2/1) mixed solvent and mixed. To the 20% adhesive solution thus obtained were added zomepirae sodium and citric acid fire rubber of the contents of zomepirae sodium and citric acid after drying were 10 wt% and 2 wt%, respectively. The zer sulting mixture was coated on a releasing liner made of a polyester in such an amount that drug content was 400 udc/m², and then dried to prepare a pressure-sensitive adhesive material laver.

This pressure-sensitive adhesive material layer was transferred to a polyethylene film having a thickness of 30 µm to produce an anti-inflammatory analgesic adhesive preparation of the present invention.

Example 7

100 Parts of isoprene rubber (molecular weight: 840,000), 30 parts of polybutene (molecular weight: 1,260) and 30 parts of an alicyclic saturated hydrocarbon resin (molecular weight: about 700; melting point: 100°C) were dissolved in toluene and mixed.

To the 20 wt%, adhesive solution thus obtained were added aluminum flufenamate and salicylic acid in such amounts that the contents of aluminum flufenamate and salicylic acid after drying were 20 wt% and 10 wt% based on the pressure-sensitive adhesive material layer, respectively, and the resulting mixture was coated on an ethylene-vinyl acetate copolymer film (vinyl acetate content: 19 wt%) having a thickness of 30 µm in such an amount that the drug content was 800 µg/cm² to form a pressure-sensitive adhesive material layer, thereby preparing an anti-inflammatory analgesic adhesive preparation of the present invention.

Comparative Examples 1 to 5

The anti-inflammatory analgesic adhesive preparation was prepared in the same method as in 4s Examples 1, 3, 5, 6 and 7 (which correspond Comparative Examples 1 to 5, respectively) except that citric acid, succinic acid, tartaric acid or salicylic acid as the organic acid was not used.

Test Example 1

Using the anti-inflammatory analgesic adhesive preparations obtained in each of Examples and 50 Comparative Examples, the inhibition effect of carrageenin foot edema was measured.

The results obtained are shown in Table 1 below.

TABLE 1

5		Volume of foot edema±S.D.	Inhibition ratio of edema (%)
-	No treatment (control)	1.25±0.13	-
10	Example 1	0.45±0.09	64.0
	Example 2	0.46±0.15	63.2
	Example 3	0.51±0.11	59.2
15	Example 4	0.39±0.10	68.8
	Example 5	0.56±0.13	55.2
20	Example 6	0.58±0.18	53.6
	Example 7	0.49±0.08	60.8
	Comparative Example 1	0.80±0.16	36.0
	Comparative Example 2	0.89±0.19	28.8
	Comparative Example 3	0.83±0.13	33.6
30	Comparative Example 4	0.94±0.20	24.8
	Comparative Example 5	0.86±0.13	31.2

Test method

WS rats (weight: about 180 g) were used. The number of animals was that each group consisted of 10 rats.

The volume of the right hind foot of each rat was measured, and a sample path (1×2 cm) was applied onto the right hind footpad. After 2 hours, the sample was removed, and 0.05 ml of a 0.5% solution of carrageenin in physiological saline was subcutaneously injected in the same right hind footpad. Three hours after the injection, the volume of the right hind foot was measured. The difference in the volume of the right hind foot was measured. The difference in the volume of the right hind foot between before and after applying of the sample patch was defined as a volume of foot edems.

The inhibition ratio of carrageenin foot edema was calculated by the following equation.

Inhibition ratio of foot edema=
$$\frac{Vc-Vt}{Vc}$$
×100

wherein Vc represents the average volume of foot edema in a control group, and Vt represents the average volume of foot edema in the group in which the test sample path was applied.

Test Example 2

The transfer percentage and the transfer amount of the anti-inflammatory analgesic agent when the same adhesive preparations as used in Test Example 1 each was applied to the skin of a human body were measured.

55 The results obtained are shown in Table 2 below.

Test method

A test sample (3'4.45 cm) was applied to the back of a human body for 24 hours and then peeled off. The residual anti-inflammatory analogesic agent was extracted with methanol, the transfer percentage and the transfer amount of the agent to the skin surface were calculated from the initial content. Each value in Table 2 is an average value of five subject.

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TABLE 2

		Transfer percentage %	Transfer amount (µg/cm²)
•	Example 1	15.9	66.8
	Example 2	15.2	62.3
,	Example 3	9.5	40.5
	Example 4	9.9	78.2
ī	Example 5	11.6	69.2
	Example 6	8.3	35.0
	Example 7	11.4	90.1
,	Comparative Example 1	3.2	13.4
	Comparative Example 2	2.6	11.1
5	Comparative Example 3	3.1	18.6
	Comparative Example 4	1.4	5.9
	Comparative Example 5	2.9	22.9
,			

It can be seen from the results shown in Tables 1 and 2 that the adhesive preparation of the present invention can provide higher anti-inflammatory effect and greater drug transfer amount of the human skin as compared to the Comparative Examples. Therefore, the adhesive preparation of the present invention is effective in the treatment of diseases.

Claims

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- An anti-inflammatory analgesic adhesive preparation comprising a flexible support having laminated thereon a pressure-sensitive adhesive layer which contains (i) a non-steroidal anti-inflammatory analgesic agent in a salt form, other than Dichlofenac Sodium, and (iii) an organic acid.
- 2. A preparation as claimed in Claim 1, wherein the non-steroidal anti-inflammatory analgesic agent having a salt form is at least one of tolmetin sodium, fenoprofen calcium, addium meclofenamate, amfenac sodium, zomepirae sodium, loxoprofen sodium and aluminum flufenamate.
- 3. A preparation as claimed in Claim 1 or 2, wherein the organic acid is a carboxylic acid.
- 4. A preparation as claimed in Claim 3, wherein the carboxylic acid is selected from citric acid, succinic acid, strataric acid, maleic acid, fumaric acid, salicylic acid and acetic acid.
 5. A preparation as claimed in any preceding claim, wherein the amount of the organic acid is 5 to 100
- parts by weight per 100 parts by weight of the anti-inflammatory analgesic agent.

 6. A preparation as claimed in Claim 5, wherein the amount of the organic acid is 10 to 50 parts by
- weight per 100 parts by weight of the anti-inflammatory analgesic agent.

 7. A preparation as claimed in any preceding claim, wherein the amount of the anti-inflammatory
- 7. A preparation as claimed in any preceding claim, wherein the amount of the anti-inflammatory analgesic agent is 1 to 40% by weight based on the weight of the pressure-sensitive adhesive material.
- 8. A preparation as claimed in Claim 7, wherein said amount of the analgesic agent is 5 to 30% by weight of the pressure-sensitive adhesive material.
 - A preparation as claimed in any preceding claim, wherein the amount of the anti-inflammatory analgesic agent is 20 to 1,800 µg/cm² per unit area.
 A preparation as claimed in Claim 9, wherein the amount of the analgesic agent is 100 to 1,200
- μg/cm².

 11. A preparation as claimed in any preceding claim, wherein the pressure-sensitive adhesive material is an acrylic pressure-sensitive adhesive material.
- 12. A preparation as claimed in Claim 11, wherein the acrylic pressure-sensitive adhesive material is a copolymer of an alkyl ester of acrylic or methacrylic acid, and/or an alkyl ester of acrylic or methacrylic acid containing an ether bond in the molecule, and another copolymerizable monomer.
 - 13. A preparation as claimed in Claim 12, wherein the proportion of the alkyl ester of (meth)acrylic

acid/the alkyl ester of (meth)acrylic acid containing an ether bond in the molecule/the other copolymerizable monomer is 40 to 80/59 to 10/1 to 40.

Patentansprüche

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- Entzündungstemmende, analgeische Klebstoffzubereitung, umfassend einen flexiblen Träger, welchem eine druckempfindliche Klebstoffzubereitung aufgelebt ist, welche (i) ein von Dichlofenantrium verschiedenes, nicht-steroides, entzündungshemmendes, analgetisches Mittel und (ii) eine organische Säure enthält.
- 2. Zubereitung gemäss Anspruch 1, worin das nicht-steroide, entzündungshemmende, analgetische Mittel in Salzform mindestens eines der folgenden ist: Tollmetin-natrium, Fenoprofenkalzium, Natriummeclofenamat, Amfena-chartium, Zomepirac-natrium, Loxoprofen-natrium und Aluminium-flufenamat.
 - Zubereitung nach Anspruch 1 oder 2, worin die organische Säure eine Carbonsäure ist.
 Zubereitung gemäss Anspruch 3, worin die Carbonsäure ausgewählt ist aus Zitronensäure,
- 4. Zubereitung gemäss Anspruch 3, worin die Carbonsäure ausgewählt ist aus Zitronensäure, 15 Bernsteinsäure, Meinsäure, Maleinsäure, Fumarsäure, Salicylsäure und Essigsäure.
 5. Zubereitung gemäss einem der vorhergehenden Ansprüche, worin die Mende der organischen
 - 5. Zubereitung gemass einem der vornergehenden Anspruche, worm die Menge der organischen Säure 5 bis 100 Gew.-Teile pro 100 Gew.-Teile des entzündungshemmenden, analgetischen Mittels ist. G. Zubereitung gemäss Anspruch 5, worin die Menge der organischen Säure 10 bis 50 Gew.-% pro 100 Gew.-% des entzündungshemmenden, analgetischen Mittels ist.
 - Zubereitung gemäss einem der vorhergehenden Ansprüche, worin die Menge des entzündungshemmenden, analgetischen Mittels 1 bis 40 Gew.-% auf Basis des druckempfindlichen Klebstoffmaterials herrägt
 - Zubereitung gemäss Anspruch 7, worin die genannte Menge des analgetischen Mittels 5 bis 30 Gew.-% des druckempfindlichen Klebstoffmaterials beträgt.
 - 9. Zubereitung gemäss einem vorhergehenden Ansprüche, worin die Menge des entzündungshemmenden, analgetischen Mittels 20 bis 1600 hg/cm² pro Flächeneinheit beträgt.
 10. Zubereitung gemäss Anspruch 9, worin die Menge des analgetischen Mittels 100 bis 1200 µg/cm²
 - Zubereitung gemäss Anspruch 9, worin die Menge des analgetischen Mittels 100 bis 1200 μg/cm² beträgt.
- Zubereitung gemäss einem der vorhergehenden Ansprüche, worin das druckempfindliche Klebstoffmaterial ein druckempfindliches Acryl-Klebstoffmaterial ist.
 Zubereitung gemäss Anspruch 11, worin das druckempfindlichen Acryl-Klebstoffmaterial ein
 - Z. Zuoteriumig ugertass Arispiruch II, worln das druckempfnolitien Archiv-Nebstormaterial ein Copolymer eines Alkylesters von Acryl- oder Methacrylsäure und/oder eines Alkylesters von Acryl- oder Methacrylsäure, die eine Etherbindung im Molekül enthält, oder eines anderen copolymerisierbaren Monomers ist.
 - 13. Zubereitung gemäss Anspruch 12, worin der Teil des Alkylesters von (Meth)acrylsäure/des Alkylesters von (Meth)acrylsäure, die eine Etherbindung im Molekül enthält/des anderen copolymeriserbaren Monomers 40 bis 80/50 bis 10/1 bis 40 beträdt.

Revendications

- 1. Préparation adhésive analgésique anti-inflammatoire comprenant un support flexible sur lequel est laminée une couche auto-adhésive, contenant (i) un agent analgésique anti-inflammatoire non-stéroide sous forme de sel, autre que le Dichlofénac de sodium et (iii) un acide organique.
- 2. Préparation selon la revendication 1, dans laquelle l'agent analgésique anti-inflammatoire onn-steroide sous forme de sel est au moins un sel parmi le tolmétine de sodium, le frénporfène de calcium, le moclofénamate de sodium, l'amfénac de sodium, le zomepirac de sodium, le loxoprofène de sodium et le flutfénamate d'aluminium.
 - 3. Préparation selon la revendication 1 ou 2, dans laquelle l'acide organique est un acide carboxylique.
- a. 4. Préparation selon la revendication 3, dans laquelle l'acide carboxylique est sélectionné parmi l'acide citrique, l'acide succinique, l'acide tartrique, l'acide maléique, l'acide fumarique, l'acide salicylique et l'acide acédique.
- 5. Préparation selon l'une quelconque des revendications précédentes, dans laquelle la quantité d'acide organique est de 5 à 100 parties en poids pour 100 parties en poids de l'analgésique sa anti-inflammatoire.
 - 6. Préparation selon la revendication 5, dans laquelle la quantité de l'acide organique est de 10 à 50 parties en poids pour 100 parties en poids de l'analgésique anti-inflammatoire.
 - 7. Préparation selon l'une quelconque des revendications précédentes, dans laquelle la quantité de l'analgésique anti-inflammatoire est de 1 à 40% en poids par rapport au polids de la matière auto-adhésive.
 8. Préparation selon la revendication 7, dans laquelle la dite quantité d'analgèsique est de 5 à 30% en
 - poids de la matière auto-adhésive. 9. Préparation selon l'une quelconque des revendications précédentes, dans laquelle la quantité d'analgésique anti-infiammatoire est de 20 à 1,600 µg/cm² par unité de surface.
- 10. Préparation selon la revendication 9, dans laquelle la quantité de l'analgésique est de 100 à 1,200 μg/cm².

- 11. Préparation selon l'une quelconque des revendications précédentes dans lequel la matière auto-adhésive est une substance auto-adhésive acrylique.
- 12. Préparation selon la revendication 11, dans laquelle la matière auto-adhésive arrylique est copolymère d'un ester d'alkyle d'acide avrylique ou méthacrylique d'un ester d'alkyle d'acide avrylique ou méthacrylique contenant une lieison éther dans la molécule, et un autre monomère copolymériseble.
 - 13. Préparation selon la revendication 12, dans laquelle la proportion d'ester d'alkyl d'acide (méth)acylique d'ester d'alkyle d'acide (méth)acylique contenant un pont éther dans la molécule/de l'autre monomère copolymérisable est de 40 à 80/59 à 10/1 à 40.

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